

STRUCTURAL PHASE TRANSITION IN $y_{Ba_2}cu_{30_{7-\delta}}$: The role of dimensionality for high temperature superconductivity

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We have performed detailed high temperature X-ray diffraction to study the nature of the structural phase of $YBa_2Cu_3O_{7-\delta}$. The results indicate the existence of a reversible orthorhombic to tetragonal phase transition at a temperature close to 750°C. If the high temperature tetragonal phase is quenched-in at low temperatures the critical superconducting temperature is considerably reduced from 92.5 K. This suggests that the one dimensional Cu-O chains present in the orthorhombic structure are necessary for high temperature superconductivity.

The relationship between the structure and high temperature superconductivity in the Y-Ba-Cu-O system is an issue which relates strongly to the origin of the superconducting mechanism in the recently discovered high temperature superconductors.¹⁻² It is by now established that the superconductivity in this system originates from a compound with a (1:2:3) Y:Ba:Cu stoichiometry. $^{3-5}$ A single crystal diffraction study by Hazen et al.⁶ and Le Page et al.⁷ found a tetragonal structure based on a stacking of 3 perovskite unit cells. However, X-ray powder diffraction investigations indicate an orthorhombically distorted variant of this tetragonal cell.^{5,8} None of the above X-ray studies could conclusively determine the positions or occupation of the oxygen atoms. Recent powder neutron diffraction investigations^{9,10} have determined the structure to be orthorhombic Pmmm. The structure consists of three distorted perovskite building blocks, which contain Cu atoms at the corners, 0 atoms on the cube edges between the Cu atoms and ordered metal ions (Ba

and Y) in the center of the cube. The Y plane is totally devoid of oxygen, the Cu-O planes adjacent to the Y-planes are dimpled with the O atoms pulled towards the Y, whereas the Cu-O plane located between Ba-O planes exhibits ordered O vacancies which results in one dimensional Cu-O chains along the b-axis. Our studies, and the conflicting structural reports in the literature, lead us to believe that there might be a high-temperature structural phase transition in this compound.

In this paper we present high temperature X-ray diffraction data showing an orthorhombic to tetragonal transition as the stoichiometric compound is heated above 750 °C. If the tetragonal phase is stabilized at low temperatures by a fast quench, the sample exhibits superconductivity at a much lower temperature than the pure orthorhombic phase. This suggests that the one dimensional Cu-O chains present in the orthorhombic structure may be responsible for the superconductivity. If this hypothesis is correct it has important implications for the type of mechanisms which are responsible for superconductivity in high temperature oxides.

Stoichiometric samples with a (1:2:3)Y:Ba:Cu composition ratio were prepared using standard powder metallurgical techniques. After thorough mixing and grinding of Y₂O₃(99.999%), CuO(99.99%) and BaCO₃(99.99%) powders the

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samples were sintered in a pure, flowing oxygen atmosphere for 10 hours at 975°C and furnace cooled. We have shown earlier that this preparation method yields an ~95% single-phase compound, which has a superconducting transition temperature of 92.5 K (transition width of 1.5 K) and an orthorhombic structure. The details of the phase diagram and the orthorhombic structure have been the subject of earlier papers.^{3,9} X-ray diffraction measurements up to 1000°C have been taken with a Rigaku DMaxII diffractometer in 1 atmosphere of flowing ${\tt O}_2$ using 2 KW of ${\tt CuK}_\alpha$ radiation. The lattice spacings were determined using Cohen's least squares fitting procedure, 11 which removes possible systematic errors introduced by the thermal expansion coefficients of the sample holder and furnace assembly.

Part of the X-ray diffraction patterns as a function of temperature are shown in Figure 1. The 750°C pattern can be indexed with a tetragonal cell with intensities which cor-



Fig. 1. Selected region of in-situ, temperature dependent, X-ray diffraction intensity for a sample heated in oxygen environment.

respond well to the structure found by Hazen et al.⁶ In both the orthorhombic and tetragonal diffraction patterns the most intense lines consist of doublets but with reversed 2:1 intensity ratios. In the case of the orthorhombic structure at 20°C, the most intense peak of each doublet is an accidental degeneracy (e.g. [103] and [110]), while in the tetragonal phase a true degeneracy leads to the intensity reversal. This is shown very clearly in Fig. 1 where a series of diffraction patterns for the (103), (013), and (110) lines are displayed for various temperatures. As the temperature is increased, from 20°C, the accidental degeneracy of the orthorhombic phase is reversed. The (103) and (013) diffraction peaks merge at the crystallographic phase change.

Figure 2 shows the temperature dependence



Fig. 2. (top): <u>a</u>- and <u>b</u>-lattice constants as a function of temperature. The data is the average of two runs on two different samples. (bottom): unit cell volume and <u>c</u>-lattice constant as a function of temperature. Note the absence of anomalies through the phase transition.

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of the lattice constants and of the unit cell volume. At around 700°C the lengths of the a and b axes converge and the structure becomes tetragonal by 750°C (Fig. 2a) At the transition the tetragonal a-axis is the average of the orthorhombic a- and b-axes. The c-axis and the volume on the other hand show no major features around this temperature (Fig. 2b). From this it appears that at the transition there is no large change in the oxygen concentration. Figure 2 only shows the heating data. On cooling, the diffraction peaks broaden considerably when passing through the phase transition. This may be due to the small grain size of the orthorhombic phase as it transforms from the tetragonal structure or a mixed orthorhombictetragonal phase region. In this temperature region, the diffraction patterns do not change with time which indicates a mixed phase region. Well below the phase transition temperature (~600°C) the diffraction peaks narrow again. The observed hysteretic behavior of the transformation indicates the possibility of a first order phase transition.

In order to study the relationship between structure and superconductivity we quenched several samples from 975°C to room temperature, thus <u>partially</u> stabilizing the high temperature tetragonal structure. This was confirmed by detailed X-ray diffraction measurements. The resistivity of all the samples was measured before and after quenching, and they all exhibit similar features. The temperature dependence of the resistivity of one of the samples is shown in Figure 3. Before quenching, the sample is



Fig. 3. Resistivity of a furnace cooled sample (A) and the same sample quenched (B).

metallic with a sharp superconducting transition temperature at 92.5 K (~ 2 K transition width).

After the quench, the resistivity is much higher and metallic at high temperature, exhibits a shallow minimum around 150 K, a small drop at approximately 92 K (T_c of the single phase orthorhombic structure) and a superconducting drop at around 50 K. Although the majority phase in this sample is tetragonal, it is not possible to state that the tetragonal phase has a 50 K superconducting transition. The 50 K transition could be due to the remaining orthorhombic material (highly disordered due to O vacancies etc.) with a lower T_c . However we can state unambiguously that the tetragonal structure has a lower transition temperature than the ordered orthorhombic phase,⁹ it may even be nonsuperconducting.

The present work together with, single crystal X-ray diffraction^{6,7} and preliminary neutron scattering¹² results for the tetragonal structure (i.e., only 2-D Cu-O sheets), have important implications for the possible superconducting mechanisms present in this class of materials. The lattice parameters of the orthorhombic and tetragonal phases are very close, so these slight changes (~0.05 Å) cannot significantly change the wave function overlaps. The main difference is that in the tetragonal phase the ordering of the oxygen vacancies in the Cu-O planes at z = 0 is nonexistent^{6,7} and therefore the one dimensional Cu-O chains are absent. As the sample is quenched to low temperature the oxygens remain randomly distributed and therefore the absence of one dimensional chains is quenched into the sample. This together with the fact that this sample exhibits absence of superconductivity at 92 K implies that the one dimensional Cu-O chains play a major role in the mechanism responsible for the high T_c in this compound. If the one dimensional chains are responsible for the superconductivity, the T_c of the quenched sample might depend very delicately on oxygen vacancies and details of the preparation method. The lack of sensitivity of the superconducting transition temperature to magnetic rare-earth substitutions (for instance Ho, Gd) $^{13-16}$ in the Y site, supports the contention that the two-dimensional Cu-O planes are not solely responsible for the superconductivity. Moreover, if the chains are responsible for the superconductivity, the Cu-O planes would screen the effect of the magnetic ions. A recent discussion¹⁷ of the formal oxidation states of the two Cu atoms in this material also concludes that the Cu-O chains are responsible for the high temperature superconductivity. This should be compared with the situation occurring in the La(Ba,Sr)Cu system in which only two dimensional Cu-O planes are present and the T_c of the material is in the 35 K region.

In summary, we have performed extensive structural studies of the orthorhombic to tetragonal phase transition in the YBaCu system. The results show a well defined transition at 750 °C which does not involve a significant volume change. Quenching experiments together with depression of superconductivity in the tetragonal phase suggests that the superconductivity in the orthorhombic phase requires the existence of the one dimensional Cu-O chains, unlike the La(Ba,Sr)Cu system where only two dimensional Cu-O planes are present. ACKNOWLEDGEMENTS --

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